










## Symbol equations

Chemical equations are not all that hard if you can see what you are doing. When the equation is drawn out, you can see how the symbols and words represent the same reaction.

Here is an example showing the reaction that happens when you strike a match head:

<b>Key</b>	 = sulfur atom	 = potassium atom
	 = carbon atom	 = aluminium atom
	 = oxygen atom	 = iron atom
	 = hydrogen atom	 = sodium atom
	 = chlorine atom	

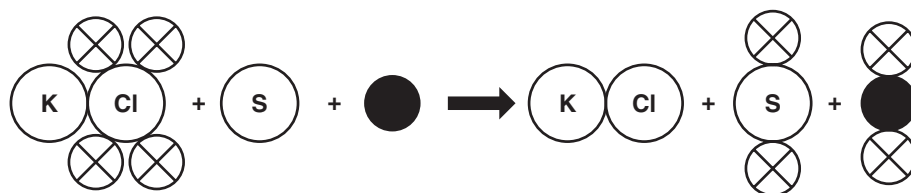
### Word equation

potassium + sulfur + carbon → potassium + sulfur + carbon  
chlorate                      chloride                      dioxide dioxide

### Symbol equation

$\text{KClO}_4 + \text{S} + \text{C} \rightarrow \text{KCl} + \text{SO}_2 + \text{CO}_2$

### Picture equation



If you follow each substance downwards in the three equations, you can see they are talking about the same reaction.

## Symbol equations (continued)

### So try this

Your teacher will give you a sheet of paper with some equations on. Complete the **symbol equation** in each of these examples. Make sure you get the numbers right in each case.

### 1 Spirit burner reaction

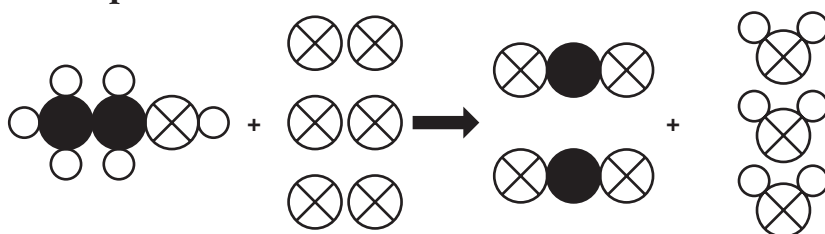
#### Word equation

ethanol + oxygen → carbon dioxide + water (vapour)

#### Symbol equation

$C_2H_5OH + \quad \rightarrow \quad + \quad$

#### Picture equation



### 2 Carbonate fizz

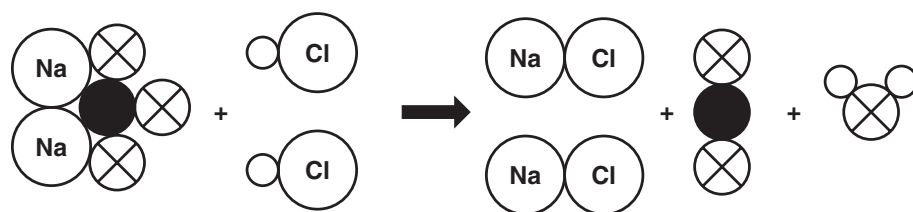
#### Word equation

sodium carbonate + hydrochloric acid → sodium chloride + carbon dioxide + water

#### Symbol equation

$Na_2CO_3 + \quad \rightarrow \quad + \quad + \quad$

#### Picture equation



### 3 The Thermit Reaction

#### Word equation

iron oxide + aluminium (metal) → aluminium oxide + iron (metal)

#### Symbol equation

$\quad + \quad \rightarrow \quad Al_2O_3 + \quad$

#### Picture equation

